Printed Page 1 of 1		Sub Code:BP104T
Paper Id: 150104	Roll No:	
,	B. PHARM CORY EXAMINATION 2019-20 ICAL INORGANIC CHEMISTE	RY
Time: 3 Hours www.a	ktupreviousyearpaper.in	Total Marks: 75
Note: 1. Attempt all Sections.		

# SECTION A

## 1. Attempt all questions in brief.

 $10 \times 2 = 20$ 

a.	Composition of barium sulphate reagent.	
b.	Give the importance of limit test.	
c.	Give the mechanism of KMnO <sub>4</sub>	
d.	Write formula of Epsom salt.	
e.	Give composition of ORS solution.	
f.	Write the method for measurement of isotonicity.	
g.	Write formula and uses of green vitriol.	
h.	Define astringents with example.	www.aktupreviousyearpaper.in
i.	Define isotope and isobar.	
j.	Differentiate between alpha, beta and gamma rays.	

#### **SECTION B**

## 2. Attempt any two parts of the following:

 $2 \times 10 = 20$ 

a.	Define expectorant; give the preparation, properties, assay and uses of ammonium chloride.
b.	Give the preparation, properties, assay and uses of Hydrogen Peroxide and Chloride lime.
c.	Write a note on fluoride in the treatment of dental caries with reference to sodium fluoride.

## **SECTION C**

## 3. Attempt any five parts of the following:

 $5 \times 7 = 35$ 

a.	Explain the methods of preparation, identification test, and test for purity of Bentonite and Aluminum hydroxide gel.
b.	Explain the properties of an ideal antacid; give the method for preparation of sodium bicarbonate and potassium permanganate.
c.	Define Haematinics. Write preparation and assay of ferrous sulfate.
d.	Write in detail the limit test for iron. www.aktupreviousyearpaper.in
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e.	Define radiopharmaceuticals and elaborate their Pharmaceutical applications.
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